A new family of fullerene derivatives: fullerencurcumin conjugates for biological and photovoltaic applications†

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The synthesis and characterization of a family of [60]fullerocurcuminoids obtained via Bingel reactions is reported. The new C60 derivatives include curcumin and curcuminoids with a variety of end groups. Preliminary biological experiments show the potential activity of the compound containing a curcumin addend, which exhibits moderate anti-HIV-1 and radical scavenger properties, but no anti-cancer activity. In addition, the new fullerocurcuminoids exhibit HOMO/LUMO energy levels that are reasonably matched with those of perovskites and when they were tested in perovskite solar cells (PSCs) as the electron transporting material (ETM), photocconversion efficiencies ranging from 14.04–14.95% were obtained, whereas a value of 16.23% was obtained for [6,6]-phenyl-C61-butyric acid methyl ester (PC61BM) based devices.

Introduction

Since the discovery of C60 in 1985 by Kroto et al.,1 fullerenes have attracted considerable attention due to their potential applications in biology and materials science.2,4 In 1993, Wudl and co-workers reported that fullerenes can inhibit HIV-1 viral infectivity, and they proposed a fulleren-induced inhibition mechanism involving binding to the protease (PR) active site based on the effect of these compounds on the in vitro activity of HIV-1 protease combined with in silico predictions.5,6 However, this paradigm was recently challenged by some of us, who showed that fullerenes do not inhibit HIV-1 protease activity in in vitro assays at doses that potently block HIV-1 maturation and infection. These findings indicate an alternative mode of viral inhibition that is yet to be determined.7,8

The potential use of fullerene C60 and its derivatives as strong antioxidants to reduce reactive oxygen species (ROS) has also been explored.9–12 Hydroxyl radicals are the most harmful among all species within the ROS family,13 and previous studies have shown that C60 and its derivatives effectively scavenge hydroxyl radicals via reactions with the double bonds.7–10,16

Curcumin (CCM) is a polyphenolic natural product found in the food coloring turmeric (curry powder), a member of the ginger family.17 CCM possesses diverse biological properties, such as anti-tumor,18 anti-oxidant,19 anti-artheritic,20 anti-amyloid,21 and anti-inflammatory,22–23 among others.24,25 These potential biomedical applications are hindered by its low bioavailability and poor water solubility, although it is readily soluble in ethanol, DMSO, and acetone.26 On the other hand, straightforward synthetic methods for low cost and high yield preparation of curcumin have been reported.27,28 These facts have motivated the design and synthesis of curcumin derivatives, called curcuminoïds (CCMoids), with the goal of overcoming some of the drawbacks, and many structural variations have been reported.29

For many years researchers have been studying the interesting biological properties of both curcumin, fullerene C60 and their derivatives as individual entities.5,29 However, to our
knowledge, no one has covalently combined these two structures into conjugated derivatives. In this work, we present a straightforward approach to synthesize several fullerocurcuminoid C₆₀ derivatives using the methylene between the two carbonyl groups to form a methanoadduct through an addition/elimination reaction (Bingel reaction).³⁰ Preliminary biological and photophysical properties of these compounds were studied, testing the activity of C₆₀-CCM against HIV, as a radical scavenger and for anticancer action. Some of the new compounds were also tested in inverted PSCs as ETMs. The main focus of this work is to provide a methodology for the synthesis of C₆₀-CCMoids and to study some potential applications; thus detailed application studies are beyond the scope of this work.

Results and discussion

Synthesis of the fullerocurcuminoid C₆₀ derivatives 2a–k

One of the most versatile reactions to functionalize fullerenes is the Bingel reaction, which involves an addition–elimination process to yield methanofullerenes.³⁰ The target compounds 2a–k were synthesized using this reaction, where the Br-CCMoid anions were generated in situ using carbon tetrabromide (CBr₄) and 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) in ortho-dichlorobenzene (o-DCB) at room temperature. As represented in Scheme 1 eleven CCMoids 1a–k were reacted with C₆₀ in the presence of CBr₄ and DBU to yield the fullerocurcuminoid derivatives 2a–k in 23–55% yields, using 3–30 min reaction times (see the ESI† for details). Longer reaction times led to the formation of bis- and poly-adduct products, which were not

Fig. 1 Analysis of the viral life cycle step affected by compound 2a. (a) Effects on the early stages of the HIV-1 viral life cycle. (b) Effects of compound 2a and indinavir on the late phase of the HIV-1 infection. VSV-G pseudotyped, single-round HIV-1 expressing luciferase were produced in the presence of DMSO, compound 2a (3 μM) and indinavir (0.1 μM); the last one is a protease inhibitor used as a control.
isolated. Among all the new derivatives 2k was the least soluble and the most unstable derivative; after one-week several dark spots were observed by thin layer chromatography (TLC). The chemical structures of compounds 1b-k were confirmed by 1H- and 13C-NMR spectroscopy (Fig. S1†) and 2a-k were characterized by matrix-assisted-laser-desorption-ionization-mass-spectrometry (MALDI-TOF-MS), 1H- and 13C-NMR spectroscopy and UV-vis absorption (see ESI, Fig. S11–S12†).

The synthesized fullerocurcuminoids exhibited poor solubility in polar solvents such as DMSO or water, except for compound 2a, due to the high solubility of the pristine CCM addend in DMSO. This low solubility in polar solvents and/or aggregation phenomena limited the biological studies exclusively to compound 2a. Compound 2a was tested as an anti-HIV-1 inhibitor of the viral replication in the early and late stages, as a radical scavenger, and as an anticancer agent against triple negative breast cancer (MDA-MB-231) and acute lymphoblastic leukemia (CEM) cells.

Effect of compound 2a on the early stages of the HIV-1 life cycle

The effect of compound 2a on the infectivity of VSV-G pseudotyped HIV-1 single-round infection viruses expressing LTR-driven luciferase (HIV-luc) was evaluated using the human CD4+ T cell line SupT1. These cells were infected with HIV-luc in the presence of DMSO (vehicle control) or fullerene 2a (10 μM) and 24 h later the compound and the input virus were removed. After three days cellular luciferase and ATP levels were measured, and luciferase was normalized to ATP to adjust for cell viability and number. In these experiments, compound 2a only minimally affected HIV-1 infectivity (Fig. 1a). These results indicate that fullerene 2a did not affect the early stages of the HIV-1 life cycle required for HIV-driven luciferase expression including viral entry and uncoating, reverse transcription, integration of viral cDNA, and HIV-1 Tat-mediated gene expression. These results also demonstrated that these compounds were not toxic to SupT1 cells at 10 μM concentration.

Effect of compound 2a on the late stages of the HIV-1 life cycle

To evaluate the effects of 2a in the late stages of viral development, we produced HIV-luc by plasmid transfection in HEK293T cells (obtained from Dr Eric Poeschla lab from Mayo Clinic) in the presence of DMSO, fullerene 2a (3 μM), or indinavir (0.1 μM). The latter is a PR inhibitor known to potently inhibit HIV-1 maturation. Viruses were concentrated by ultracentrifugation through a sucrose cushion, normalized for p24 levels, and their infectivity evaluated in untreated SupT1 cells (Fig. 1b). As expected, indinavir resulted in 96% inhibition at 0.1 μM, as compared to DMSO. On the other hand, compound 2a inhibited by 64% the infectivity at 3 μM (Fig. 1b).

The plasmids used to generate retroviral vectors were described previously.23 HIV-1-derived vectors were produced using pHIV Luc and pMD.G. pHIV Luc was derived from pNL4-3.Luc.R-E.23 by introducing a deletion in the env open reading frame (Bright-Glow™ Luciferase Assay System, Promega).

Compound 2a exhibits cytoprotective activity in human neuroblastoma cells undergoing oxidative stress

A human SH-SY5Y cell line (ATCC) was selected to review the cytoprotective activity of compound 2a; an established cell
model of dopaminergic neurons to study Parkinson’s disease in vitro. The cell death percentages were quantified by using the DNS assay and examined via high throughput screening bioimager system (GE Healthcare). Cells pretreated (1 h) with 0.3, 3, 10 and 30 μM of compound 2a were subsequently exposed (24 h) to 100 μM of H2O2, to inflict the cellular oxidative stress and cell death. As expected, the H2O2-treated cells, without any pretreatment, exhibited the maximum value of cytotoxicity (21.71 ± 6.22%; Fig. 2). All concentrations of compound 2a tested displayed a consistently significant reduction in cytotoxicity after adding H2O2 (100 μM; 24 h) when compared with the cells treated with just H2O2 (Fig. 2). The cytotoxicity percentage values were 5.43 ± 1.89, 3.76 ± 1.79, 4.69 ± 2.41 and 4.92 ± 1.9 (average ± SD), when 0.3, 3, 10 and 30 μM of compound 2a were tested, respectively (Fig. 2); around 77% in reduction of cytotoxicity, which is considered highly significant (P < 0.00001). Moreover, compound 2a itself showed no cytotoxic activity at all concentrations reviewed after 25 h of cell exposure (around 4.7%; light gray bars; Fig. 2). These cytotoxicity percentage values were very similar to DMSO and untreated negative control cells. These findings indicate that compound 2a alone is not toxic to cells under the conditions tested in this study. Therefore, even at nanomolar concentrations (300 nM), compound 2a possesses a strong antioxidant and neuroprotective activity on human dopaminergic neurons SH-SY5Y cells, experiencing H2O2-induced oxidative stress, probably acting as an effective free radical scavenger reagent.

**Effect of compound 2a as an anticancer agent**

The potential anticancer activity of compound 2a was examined in two human cancer cell lines; MDA-MB-231 triple negative breast cancer and CEM acute lymphoblastic leukemia cells (both from ATCC, Manassas, VA) by a differential nuclear staining assay (see ESI† for details). Both cell lines exhibited insignificant amounts of cytotoxicity when exposed to compound 2a in a gradient of concentrations ranging from 1 to 10 μM for 48 h (Fig. 3). As shown in Fig. 3 the cytotoxicity values observed for compound 2a-treated cells lines consistently resembled the values of the untreated (Unt) and DMSO vehicle controls, suggesting that compound 2a has no cytotoxic activity against triple negative breast cancer and acute lymphoblastic leukemia cells. H2O2 (1 mM) was included as a positive control of death, and as expected, the highest cytotoxic values were observed; 22.9% and 76.1% for MDA-MB-231 and CEM cells respectively. These last results indicate that MDA-MB-231 cells were more resistant to the H2O2 (1 mM), as compared with CEM cells.

**Photovoltaic properties of fullerocuruminoids 2a–h**

Due to the extended conjugation of the CCM/CCMoid backbones, these compounds have been extensively studied as light-harvesting materials in bulk heterojunction (BHJ) and dye-sensitized (DS) solar cells, and PCE values up to 4.1% and 5.8% have been obtained, respectively. Since fullerocuruminoids 2a–h have not been synthetized before they have not been tested in PSCs as the electron transporting layers (ETLs) nor as the hole transporting layers (HTLs).

The optical properties of the fullerocuruminoid derivatives 2a–k were investigated by means of UV/Vis spectroscopy. As seen in Fig. S3,† 2a–k exhibit absorption bands below 400 nm arising from both functionalities, C=C and the aromatic endings of the CCMoids. Between 400–450 nm the bands observed for the majority of the compounds correspond to the skeleton of the CCMoid moieties and in the case of 2e, the absorption band at the highest wavelength is linked to the ferrocene groups.

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**Table 1**  Electrochemical and photophysical data of compounds 2a–h.

<table>
<thead>
<tr>
<th>Compound</th>
<th>λ&lt;sub&gt;abs&lt;/sub&gt; (nm)</th>
<th>E&lt;sub&gt;g&lt;/sub&gt; (eV)</th>
<th>E&lt;sub&gt;onset&lt;/sub&gt; red (V)</th>
<th>LUMO (eV)</th>
<th>HOMO (eV)</th>
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<tr>
<td>PC&lt;sub&gt;61&lt;/sub&gt;BM</td>
<td>718</td>
<td>1.73</td>
<td>−0.90</td>
<td>−3.90</td>
<td>−5.63</td>
</tr>
<tr>
<td>2a</td>
<td>716</td>
<td>1.73</td>
<td>−1.09</td>
<td>−3.93</td>
<td>−5.66</td>
</tr>
<tr>
<td>2b</td>
<td>718</td>
<td>1.73</td>
<td>−1.08</td>
<td>−3.94</td>
<td>−5.67</td>
</tr>
<tr>
<td>2c</td>
<td>718</td>
<td>1.73</td>
<td>−1.09</td>
<td>−3.93</td>
<td>−5.66</td>
</tr>
<tr>
<td>2d</td>
<td>714</td>
<td>1.74</td>
<td>−1.13</td>
<td>−3.89</td>
<td>−5.63</td>
</tr>
<tr>
<td>2e</td>
<td>720</td>
<td>1.72</td>
<td>−1.07</td>
<td>−3.95</td>
<td>−5.67</td>
</tr>
<tr>
<td>2f</td>
<td>716</td>
<td>1.73</td>
<td>−1.02</td>
<td>−4.00</td>
<td>−5.73</td>
</tr>
<tr>
<td>2g</td>
<td>718</td>
<td>1.73</td>
<td>−1.10</td>
<td>−3.92</td>
<td>−5.65</td>
</tr>
<tr>
<td>2h</td>
<td>718</td>
<td>1.73</td>
<td>−1.09</td>
<td>−3.93</td>
<td>−5.66</td>
</tr>
</tbody>
</table>

*Values obtained using the following formula*  

E<sub>LUMO</sub> = −(E<sub>onset</sub> red − Fe<sup>2+</sup>/Fe<sup>3+</sup> + 4.8) eV.†

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**Fig. 4**  (a) Schematic illustration of the estimated HOMO and LUMO energy levels, calculated from CV and UV-vis.  (b) PSC’s cell architectures.
The electrochemical properties of 2a–h and PC$_{61}$BM were determined by cyclic voltammetry (CV) in o-DCB as described in the EIS. As shown in Fig. S34,† compounds 2a–h exhibit reversible and quasi-reversible cathodic electrochemical behavior between −1.0 and −1.3 V, whereas more negative potentials lead to irreversible retro-cycloadditions. In particular compounds 2d, 2e and 2f showed chemical reversibility for the first reduction step at a scan rate of 100 mV s$^{-1}$, followed by chemical and electrochemically irreversible waves. The highest occupied molecular orbital/lowest unoccupied molecular orbital (HOMO/LUMO) values were estimated from the UV and CV measurements. The optical properties of compounds 2a–h are summarized in Table 1.

Compounds 2a–h exhibit similar HOMO/LUMO values to those of PC$_{61}$BM, thus they are a good match for the conduction and valence bands of the perovskite, respectively, and should provide efficient electron collection and hole blocking ability. The schematic representation of the work function of the PSCs is illustrated in Fig. 4a.

We studied the electron transporting ability of compounds 2a–h in PSCs with the configuration of ITO/PEDOT:PSS/perovskite/ETM/Ag (Fig. 4b), and compared them to PC$_{61}$BM. The cells were fully characterized using SEM, XRD, and photovoltaic measurements (Fig. S35 and S36,† respectively). As shown in Fig. 5a, devices based on 2a–h show PCEs ranging from 14.04–14.95%. The main photovoltaic characteristics are summarized in Table 1.

Table 2  Summary of the main photovoltaic performance of PSCs using 2a–h as the ETMs$^a$

<table>
<thead>
<tr>
<th>Compound</th>
<th>Calculated $J_{sc}$ (mA cm$^{-2}$)</th>
<th>$J_{sc}$ (mA cm$^{-2}$)</th>
<th>$V_{oc}$ (V)</th>
<th>FF (%)</th>
<th>PCE (%)</th>
</tr>
</thead>
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<tr>
<td>PC$_{61}$BM</td>
<td>21.87</td>
<td>22.10</td>
<td>0.93</td>
<td>0.79</td>
<td>15.73 ± 0.46 (16.23)</td>
</tr>
<tr>
<td>2a</td>
<td>21.02</td>
<td>21.14</td>
<td>0.89</td>
<td>0.76</td>
<td>13.92 ± 0.38 (14.30)</td>
</tr>
<tr>
<td>2b</td>
<td>20.54</td>
<td>20.79</td>
<td>0.90</td>
<td>0.77</td>
<td>13.79 ± 0.62 (14.41)</td>
</tr>
<tr>
<td>2c</td>
<td>20.35</td>
<td>20.77</td>
<td>0.89</td>
<td>0.76</td>
<td>13.69 ± 0.35 (14.04)</td>
</tr>
<tr>
<td>2d</td>
<td>20.63</td>
<td>20.87</td>
<td>0.89</td>
<td>0.76</td>
<td>13.44 ± 0.68 (14.12)</td>
</tr>
<tr>
<td>2e</td>
<td>21.27</td>
<td>21.33</td>
<td>0.90</td>
<td>0.77</td>
<td>13.47 ± 1.31 (14.78)</td>
</tr>
<tr>
<td>2f</td>
<td>21.09</td>
<td>21.33</td>
<td>0.91</td>
<td>0.77</td>
<td>14.46 ± 0.49 (14.95)</td>
</tr>
<tr>
<td>2g</td>
<td>21.16</td>
<td>21.25</td>
<td>0.90</td>
<td>0.76</td>
<td>14.20 ± 0.34 (14.54)</td>
</tr>
<tr>
<td>2h</td>
<td>20.86</td>
<td>21.02</td>
<td>0.89</td>
<td>0.77</td>
<td>13.74 ± 0.66 (14.40)</td>
</tr>
</tbody>
</table>

$^a$ Calculated $J_{sc}$ from EQE measurements, values in parentheses are the highest PCEs.
summarized in Table 2. These PCEs, although lower than for the control PC61BM (16.23%), are significant and promising. The integrated photocurrent densities based on EQE measurements (Fig. 5b) are consistent with those obtained from J-V measurements (Table 2). Devices prepared with all the fullerene derivatives including PC61BM exhibit a high photo response within the entire visible range. Device performance reproducibility was calculated from the PCE distributions measured for 32 independent cells (Fig. 5c). The long term-device stabilities for PSCs were conducted under ~20% humidity at room temperature without encapsulation for seven days. Fig. 5d, shows the normalized PCEs vs. time. After seven days of periodic measurements, devices based on 2a–h and PC61BM lost 60% of their initial PCE. No significant differences were observed for devices prepared either with compounds 2a–h or PC61BM. These encouraging results will be studied in more detail in the future to fully understand the roles of aggregation as well as that of interface interactions between the C60-CCMoids and the perovskite films.

Conclusions

In this work, we describe for the first time the conjugation of curcumin-like compounds to fullerene C60 through a Bingel type reaction to obtain new fullerocurcuminoid derivatives. Fullerocurcuminoid 2a was soluble in DMSO and was tested as an anti-HIV-1 agent, as a radical scavenger, and as an anticancer agent. When studying the late stages of the HIV viral cycle, compound 2a inhibits 64% of the HIV-1 viral infection at 3 μM and it also prevents radical damage at this concentration. No anticancer activity was detected in our assays. These results are consistent with those for curcuminoid, which exhibits low toxicity and no anticancer activity toward several cell lines. In addition, fullerocurcuminoids 2a–h were soluble in chlorobenzene, and exhibit HOMO/LUMO energy levels that match those of perovskites, therefore, when they were tested in PSCs as the ETMs, PCEs in the range of 14.04–14.95%, were obtained. A value of 16.23% was obtained for PC61BM-based devices. Further studies are needed to understand the nature of the photovoltaic activity as well as to improve the observed performances.

Conflicts of interest

There are no conflicts to declare.

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